Reactions of 2-Pentynyl Trimethylsilylethynyl Selenides with Primary Amines via Allenyl Selenoketene<br>Mamoru Koketsu [a],* Masanori Kanoh [b] and Hideharu Ishihara [b]*<br>[a] Division of Instrumental Analysis, Life Science Research Center, Gifu University, Gifu 501-1193, Japan, [b] Department of Chemistry, Faculty of Engineering, Gifu University, Gifu 501-1193, Japan Received June 9, 2004

Reactions of 2-pentynyl trimethylsilylethynyl selenide $\mathbf{3}$ with primary amines $\mathbf{5}$ via allenyl selenoketene $\mathbf{4}$ afforded the corresponding selenine $\mathbf{6}$, selenophene 7 and $\beta, \gamma$-unsaturated selenoamide 8 , respectively.
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The selenoketene has been extensively investigated because of the interesting properties and the reactivity has been reported using theoretical calculations [1]. From those results, the selenoketene is found to be more reactive than ketene [2]. Recently, we confirmed interesting evidence for allenyl selenoketene formation [3] via a $[3+3]$ sigmatropic rearrangement by trapping with primary amines and its cyclization [4]. Herein, we investigated interesting reactivities of allenyl selenoketene.

The reaction leading to 2 -pentynyl trimethylsilylethynyl selenides $\mathbf{3}$ is described. 2-Pentynyl chloride 2 was added to tetrahydrofuran solution of the lithium alkyneselenoate, generated in situ from trimethylsilylacetylene $\mathbf{1}, n-\mathrm{BuLi}$ and elementary selenium, and the mixture was stirred at room temperature for 1.5 hours. Subsequent silica gel flash column chromatography afforded 2-pentynyl trimethylsilylethynyl selenide $\mathbf{3}$ in a 91 \% yield.
In the case of allenylketene, it could be isolated by the ketene-stabilizing effect of silyl substitution [2f,5]. Similarly, though we tried to isolate the present allenyl selenoketene bearing silyl group, it failed [4,6]. In order to confirm the evidence for allenyl selenoketene formation via a $[3+3]$ sigmatropic rearrangement, reactions with amines were carried out. The mixture of 3 and $n$ butyl amine 5a in benzene was refluxed for 5 hours. After standard workup 2-butylimino-4-ethyl-2H-5,6dihydroselenine 6a was isolated in a $53 \%$ yield as a major product. 2-Butylimino-4-ethyl-5-methylidene-2,5-dihydroselenophene 7a and $N$-butyl-3-ethyl-2-trimethylsilyl-3,4-pentadieneselenoamide 8a were also isolated in $5 \%$ and $21 \%$, respectively (Scheme 1). The ir and nmr spectra of 8a were typical of a terminal allene [7]. Compound $\mathbf{8}$ bore trimethylsilyl group, while the trimethylsilyl group of precursors of compounds 6 and 7 was eliminated. The reaction of 2-pentynyl phenylethynyl selenide with amine 5 gave 2-imino-4-ethyl-3-phenyl-2H-5,6-dihydroselenine [4], while the present reaction gave thermodynamically stable compounds 6 and 7 by the elimination of the trimethylsilyl group [8]. The reactions using four kinds of primary amines 5 were investigated and gave $\mathbf{6}$ in 49-59\% yields (Table 1).
Scheme 1


Table 1
Reaction of 2-Pentynyl Trimethylsilylethynyl Selenide 3 and Primary Amines 5

| Entry | Primary amine | Yield (\%) $[\mathrm{a}]$ |  |  |
| :---: | :--- | :---: | ---: | ---: |
|  | $(\mathrm{R}=)$ | $\mathbf{6}$ | $\mathbf{7}$ | $\mathbf{8}$ |
|  |  |  |  |  |
| 1 | $\mathbf{5 a}, \mathrm{CH}_{3}\left(\mathrm{CH}_{2}\right)_{3}$ | 53 | 5 | 21 |
| 2 | $\mathbf{5 b}, \mathrm{CH}_{3}\left(\mathrm{CH}_{2}\right)_{2}$ | 52 | 10 | 18 |
| 3 | $\mathbf{5 c}, \mathrm{C}_{6} \mathrm{H}_{5}\left(\mathrm{CH}_{2}\right)_{2}$ | 49 | 0 | 0 |
| 4 | $\mathbf{5 d}, \mathrm{C}_{6} \mathrm{H}_{5} \mathrm{CH}_{2}$ | 59 | 0 | 0 |

[a] Isolated yield.

The reactions with phenethyl amine 5c and benzyl amine $\mathbf{5 d}$ afforded only $\mathbf{6 c}$ and $\mathbf{6 d}$, respectively, but did not give derivatives $\mathbf{7}$ and $\mathbf{8}$. The isolation of $\mathbf{8 a}$, bearing a terminal allene, could confirm the formation of an allenyl selenoketene intermediate $\mathbf{4}$ in the present reaction and be explained by the mechanism presented in Scheme 2. Subsequent nucleophilic attack of allenyl selenoketene 4, which was generated from 3 through a [3,3] sigmatropic rearrangement, by primary amine 5 afforded the observed products 6, 7 and $\mathbf{8}$, respectively (Scheme 2 ).

Previously, a generation of allenyl selenoketene intermediate $\mathbf{4}$ by heating of selenide $\mathbf{3}$ was confirmed [4]. The present reactions could confirm the generation of the allenyl selenoketene intermediate $\mathbf{4}$ in the reaction process again. The allenyl selenoketene intermediate 4 leads to 6membered ring $2 H-5,6$-dihydroselenine 6, 5-membered ring 2 H -5-hydroselenophene 7 , and 3,4-pentadieneselenoamide $\mathbf{8}$, respectively, by the reactions with the amines.

Scheme 2


The allenyl selenoketene contains interesting and highly active sites, and gave individual products by the substituent groups or reaction conditions.

## EXPERIMENTAL

2-Pentynyl Trimethylsilylethynyl Selenide (3).
To a solution of trimethylsilylacetylene $\mathbf{1}(0.29 \mathrm{~g}, 3.0$ mmole) in dry tetrahydrofuran ( 20 ml ), $n$-butyllithium, in $n$-hexane ( $2.0 \mathrm{ml}, 3.0 \mathrm{mmole}$ ) was added and stirred at $0^{\circ} \mathrm{C}$ for 15 minutes under an argon atmosphere. Then selenium powder ( $0.24 \mathrm{~g}, 3.0 \mathrm{mmole}$ ) was added to the mixture and stirred for 30 minutes. Moreover, 2-pentynyl chloride $2(0.31 \mathrm{ml}, 3.0$ mmole) was added to the reaction mixture and stirred at room temperature for 1 hour. The mixture was extracted with diethyl ether and washed with saturated NaCl solution. The organic layer was dried over sodium sulfate and evaporated to dryness. The residue was purified by flash chromatography on silica gel with $n$-hexane: diethyl ether (50:1) to give 0.66 g 3 (91\%) as yellow green oil, ir (neat): $2089 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H} \mathrm{nmr}(400 \mathrm{MHz}$, deuteriochloroform): $\delta 0.00\left(\mathrm{~s}, 9 \mathrm{H}, \mathrm{CH}_{3}\right), 0.95(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.6$ $\mathrm{Hz}, \mathrm{CH}_{3}$ ), $2.04\left(\mathrm{qt}, 2 \mathrm{H}, \mathrm{J}=7.6,2.4 \mathrm{~Hz}, \mathrm{CH}_{2}\right), 3.35(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=$ $2.6 \mathrm{~Hz}, \mathrm{CH}_{2}$ ); ${ }^{13} \mathrm{C} \mathrm{nmr}$ ( 100 MHz , deuteriochloroform): $\delta-0.1$, $12.6,13.8,15.3,74.0,85.8,87.4,109.8 ;{ }^{77} \mathrm{Se} \mathrm{nmr}(76 \mathrm{MHz}$, deuteriochloroform): $\delta 248.6 ; \mathrm{ms}(\mathrm{EI}): \mathrm{m} / \mathrm{z}=244\left(\mathrm{M}^{+}\right)$.

General Procedure for Synthesis of 2-Alkylimino-4-ethyl-2H-5,6-dihydroselenine (6) and 2-Alkylimino-4-ethyl-5-methyli-dene-2,5-dihydroselenophene (7) and $N$-Alkyl-3-ethyl-2-trimethylsilyl-3,4-pentadiene Selenoamide (8).

To a solution of 2-pentynyl trimethylsilylethynyl selenide 3 $(0.24 \mathrm{~g}, 1.0 \mathrm{mmole})$ in dry benzene $(20 \mathrm{ml}), n$-butyl amine $\mathbf{5 a}$ $(0.20 \mathrm{ml}, 2.0 \mathrm{mmole})$ was added. The mixture was refluxed with stirring for 5 hours under an argon atmosphere. The mixture was extracted with diethyl ether and washed with saturated NaCl solution. The organic layer was dried over sodium sulfate and evaporated to dryness. The residue was purified by column chromatography on silica gel with $n$-hexane:diethyl ether (10:1) to give $6 \mathbf{a}(0.13 \mathrm{~g}, 53 \%), 7 \mathrm{a}(0.01 \mathrm{~g}, 5 \%)$ and $\mathbf{8 a}(0.07 \mathrm{~g}, 21 \%)$, respectively.

## 2-Butylimino-4-ethyl-2H-5,6-dihydroselenine (6a).

This compound was obtained as orange oil, ir (neat): 1644 $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ ( 400 MHz , deuteriochloroform): $\delta 0.95(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=$ $7.2 \mathrm{~Hz}, \mathrm{CH}_{3}$ ), $1.08\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.2 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.37-1.47(\mathrm{~m}, 2 \mathrm{H}$, $\mathrm{CH}_{2}$ ), 1.72 (quint, $2 \mathrm{H}, \mathrm{J}=7.3 \mathrm{~Hz}, \mathrm{CH}_{2}$ ), $2.16(\mathrm{q}, 2 \mathrm{H}, \mathrm{J}=7.5 \mathrm{~Hz}$, $\mathrm{CH}_{2}$ ), $2.60\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=6.0 \mathrm{~Hz}, \mathrm{CH}_{2}\right), 2.97\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=6.2 \mathrm{~Hz}, \mathrm{CH}_{2}\right)$, $3.34\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=7.2 \mathrm{~Hz}, \mathrm{CH}_{2}\right), 6.05(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}) ;{ }^{13} \mathrm{C} \mathrm{nmr}(100$ MHz , deuteriochloroform): $\delta 11.6,13.8,18.7,20.7,29.8,32.508$, $32.512,55.8,123.4,151.8,155.2 ;{ }^{77} \mathrm{Se} \mathrm{nmr}(76 \mathrm{MHz}$, deuteriochloroform): $\delta 292.3 ; \mathrm{ms}(\mathrm{CI}): \mathrm{m} / \mathrm{z}=246\left(\mathrm{M}^{+}+1\right)$.
Anal. Calcd. for $\mathrm{C}_{11} \mathrm{H}_{19} \mathrm{NSe}: \mathrm{C}, 54.09 ; \mathrm{H}, 7.84 ; \mathrm{N}, 5.73$. Found: C, 54.13; H, 7.88; N, 5.76.

4-Ethyl-2-propylimino-2H-5,6-dihydroselenine (6b).
This compound was obtained as orange oil, ir (neat): $1644 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H} \mathrm{nmr}(400 \mathrm{MHz}$, deuteriochloroform): $\delta 0.99(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.6 \mathrm{~Hz}$, $\left.\mathrm{CH}_{3}\right), 1.08\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.6 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.72-1.81\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 2.16$ $\left(\mathrm{q}, 2 \mathrm{H}, \mathrm{J}=7.6 \mathrm{~Hz}, \mathrm{CH}_{2}\right), 2.60\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=6.0 \mathrm{~Hz}, \mathrm{CH}_{2}\right), 2.97(\mathrm{t}, 2 \mathrm{H}$, $\left.\mathrm{J}=6.2 \mathrm{~Hz}, \mathrm{CH}_{2}\right), 3.30\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=7.0 \mathrm{~Hz}, \mathrm{CH}_{2}\right), 6.05(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH})$; ${ }^{13} \mathrm{C} \mathrm{nmr}$ ( 100 MHz , deuteriochloroform): $\delta 11.6,12.1,18.8,23.7$, $29.8,32.5,57.8,123.4,151.9,155.4 ;{ }^{77} \mathrm{Se} \mathrm{nmr}(76 \mathrm{MHz}$, deuteriochloroform): $\delta 292.6 ; \mathrm{ms}(\mathrm{CI}): \mathrm{m} / \mathrm{z}=232\left(\mathrm{M}^{+}+1\right)$.

Anal. Calcd for $\mathrm{C}_{10} \mathrm{H}_{17} \mathrm{NSe}: \mathrm{C}, 52.17 ; \mathrm{H}, 7.44 ; \mathrm{N}, 6.08$. Found: C, 52.33; H, 7.64; N, 5.96.

4-Ethyl-2-phenethylimino-2H-5,6-dihydroselenine (6c).
This compound was obtained as orange oil, ir (neat): 1643 $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}(400 \mathrm{MHz}$, deuteriochloroform): $\delta 1.08(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=$ $\left.7.4 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 2.16\left(\mathrm{q}, 2 \mathrm{H}, \mathrm{J}=7.5 \mathrm{~Hz}, \mathrm{CH}_{2}\right), 2.59(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=6.0$ $\left.\mathrm{Hz}, \mathrm{CH}_{2}\right), 2.96\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=6.2 \mathrm{~Hz}, \mathrm{CH}_{2}\right), 3.05(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=7.8 \mathrm{~Hz}$, $\mathrm{CH}_{2}$ ), $3.60\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=8.0 \mathrm{~Hz}, \mathrm{CH}_{2}\right), 6.06(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}), 7.15-7.34$ (m, 5H, Ar); ${ }^{13} \mathrm{C}$ nmr ( 100 MHz , deuteriochloroform): $\delta 11.6$, $18.9,29.8,32.6,36.9,57.7,123.4,126.0,128.3,128.7,140.1$, $152.3,156.3 ;{ }^{77} \mathrm{Se} \mathrm{nmr}(76 \mathrm{MHz}$, deuteriochloroform): $\delta 295.9$; $\mathrm{ms}(\mathrm{CI}): \mathrm{m} / \mathrm{z}=294\left(\mathrm{M}^{+}+1\right)$.

Anal. Calcd for $\mathrm{C}_{15} \mathrm{H}_{19} \mathrm{NSe}: \mathrm{C}, 61.64 ; \mathrm{H}, 6.55 ; \mathrm{N}, 4.79$. Found: C, 61.76; H, 6.77; N, 4.83.

## 2-Benzylimino-4-ethyl-2H-5,6-dihydroselenine (6d).

This compound was obtained as orange oil, ir (neat): 1643 $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ ( 400 MHz , deuteriochloroform): $\delta 1.08$ (t, 3H, J = $\left.7.2 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 2.16\left(\mathrm{q}, 2 \mathrm{H}, \mathrm{J}=7.3 \mathrm{~Hz}, \mathrm{CH}_{2}\right), 2.61(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=6.0$ $\left.\mathrm{Hz}, \mathrm{CH}_{2}\right), 2.99\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=6.2 \mathrm{~Hz}, \mathrm{CH}_{2}\right), 4.56\left(\mathrm{~s}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 6.14$ $(\mathrm{s}, 1 \mathrm{H}, \mathrm{CH}), 7.26-7.32(\mathrm{~m}, 5 \mathrm{H}, \mathrm{Ar}) ;{ }^{13} \mathrm{C}$ nmr $(100 \mathrm{MHz}$, deuteriochloroform): $\delta 11.6,19.0,29.7,32.5,59.7,123.6,126.7,128.0$, $128.3,139.4,152.4,156.7 ;{ }^{77} \mathrm{Se} \mathrm{nmr}(76 \mathrm{MHz}$, deuteriochloroform): $\delta 297.2 ; \mathrm{ms}(\mathrm{CI}): \mathrm{m} / \mathrm{z}=280\left(\mathrm{M}^{+}+1\right)$.

Anal. Calcd for $\mathrm{C}_{14} \mathrm{H}_{17} \mathrm{NSe}: \mathrm{C}, 60.43 ; \mathrm{H}, 6.16 ; \mathrm{N}, 5.03$. Found: C, 60.22; H, 6.23; N, 5.26.

2-Butylimino-4-ethyl-5-methylidene-2,5-dihydroselenophene (7a).

This compound was obtained as orange oil, ir (neat): 1622 $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{nmr}$ ( 400 MHz , deuteriochloroform): $\delta 0.95(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=$ $\left.7.6 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.24\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.2 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.38-1.47(\mathrm{~m}, 2 \mathrm{H}$, $\mathrm{CH}_{2}$ ), 1.72 (quint, $\left.2 \mathrm{H}, \mathrm{J}=7.3 \mathrm{~Hz}, \mathrm{CH}_{2}\right), 2.51(\mathrm{q}, 2 \mathrm{H}, \mathrm{J}=7.5 \mathrm{~Hz}$, $\left.\mathrm{CH}_{2}\right), 3.27\left(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=7.2 \mathrm{~Hz}, \mathrm{CH}_{2}\right), 5.45\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}_{2}\right), 5.89(\mathrm{~s}$, $\left.1 \mathrm{H}, \mathrm{CH}_{2}\right), 6.46(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}) ;{ }^{13} \mathrm{C} \mathrm{nmr}(100 \mathrm{MHz}$, deuteriochloroform): $\delta 12.9,13.9,20.7,21.5,32.4,61.8,112.3,133.0,143.6$, 156.7, 164.7; ${ }^{77} \mathrm{Se} \mathrm{nmr}(76 \mathrm{MHz}$, deuteriochloroform): $\delta 428.9$; $\mathrm{ms}(\mathrm{CI}): \mathrm{m} / \mathrm{z}=244\left(\mathrm{M}^{+}+1\right)$.

Anal. Calcd for $\mathrm{C}_{11} \mathrm{H}_{17} \mathrm{NSe}: \mathrm{C}, 54.54 ; \mathrm{H}, 7.07$; N, 5.78. Found: C, 54.36; H, 7.11; N, 5.92.

4-Ethyl-5-methylidene-2-propylimino-2,5-dihydroselenophene (7b).

This compound was obtained as orange oil, ir (neat): 1627 $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ ( 400 MHz , deuteriochloroform): $\delta 0.99(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=$ $\left.7.2 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.24\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.6 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.73-1.82(\mathrm{~m}, 2 \mathrm{H}$, $\left.\mathrm{CH}_{2}\right), 2.51\left(\mathrm{q}, 2 \mathrm{H}, \mathrm{J}=7.5 \mathrm{~Hz}, \mathrm{CH}_{2}\right), 3.23(\mathrm{t}, 2 \mathrm{H}, \mathrm{J}=6.8 \mathrm{~Hz}$, $\left.\mathrm{CH}_{2}\right), 5.45\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}_{2}\right), 5.89\left(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}_{2}\right), 6.46(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}) ;{ }^{13} \mathrm{C}$ $\mathrm{nmr}(100 \mathrm{MHz}$, deuteriochloroform): $\delta 12.1,12.9,21.5,23.6$, $29.7,63.9,112.3,133.0,143.6,156.8,164.7 ;{ }^{77} \mathrm{Se} \mathrm{nmr}(76 \mathrm{MHz}$, deuteriochloroform): $\delta 29.3 ; \mathrm{ms}(\mathrm{CI}): \mathrm{m} / \mathrm{z}=230\left(\mathrm{M}^{+}+1\right)$.

Anal. Calcd for $\mathrm{C}_{10} \mathrm{H}_{16} \mathrm{NSe}: \mathrm{C}, 52.40 ; \mathrm{H}, 7.04 ; \mathrm{N}, 6.11$. Found: C, 52.46; H, 7.14; N, 5.98.
$N$-Butyl-3-ethyl-2-trimethylsilyl-3,4-pentadieneselenoamide (8a).

This compound was obtained as yellow oil, ir (neat): 3326, 1947, 1525, $844 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H} \mathrm{nmr}$ ( 400 MHz , deuteriochloroform): $\delta$ $0.19\left(\mathrm{~s}, 9 \mathrm{H}, \mathrm{CH}_{3}\right), 0.97\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.4 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 0.99(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=$ $\left.7.4 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.37-1.46\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 1.63-1.72\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right)$, $1.90-2.10\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 3.52(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}), 3.61-3.80(\mathrm{~m}, 2 \mathrm{H}$, $\left.\mathrm{CH}_{2}\right), 4.99-5.10\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 8.26(\mathrm{br} \mathrm{s}, 1 \mathrm{H}, \mathrm{NH}) ;{ }^{13} \mathrm{C} \operatorname{nmr}(100$ MHz , deuteriochloroform): $\delta-1.8,11.8,13.6,20.2,28.5,30.0$, $49.4,57.4,79.6,105.1,204.6,206.9 ;{ }^{77} \mathrm{Se} \mathrm{nmr}(76 \mathrm{MHz}$, deuteriochloroform): $\delta 543.4 ; \mathrm{ms}(\mathrm{CI}): \mathrm{m} / \mathrm{z}=318\left(\mathrm{M}^{+}+1\right)$.

Anal. Calcd for $\mathrm{C}_{14} \mathrm{H}_{27} \mathrm{NSe}: \mathrm{C}, 58.32 ; \mathrm{H}, 9.44 ; \mathrm{N}, 4.86$. Found: C, 58.45; H, 9.51 ; N, 4.87.

3-Ethyl- $N$-propyl-2-trimethylsilyl-3,4-pentadieneselenoamide (8b).

This compound was obtained as yellow oil, ir (neat): 3329, 1947, 1523, $844 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \mathrm{nmr}$ ( 400 MHz , deuteriochloroform): $\delta$ $0.19\left(\mathrm{~s}, 9 \mathrm{H}, \mathrm{CH}_{3}\right), 0.99\left(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=7.0 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.01(\mathrm{t}, 3 \mathrm{H}, \mathrm{J}=$ $\left.7.0 \mathrm{~Hz}, \mathrm{CH}_{3}\right), 1.66-1.77\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 1.89-2.10\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right)$, $3.52(\mathrm{~s}, 1 \mathrm{H}, \mathrm{CH}), 3.59-3.77\left(\mathrm{~m}, 2 \mathrm{H}, \mathrm{CH}_{2}\right), 4.98-5.06(\mathrm{~m}, 2 \mathrm{H}$, $\left.\mathrm{CH}_{2}\right), 8.29$ (br s, $\left.1 \mathrm{H}, \mathrm{NH}\right) ;{ }^{13} \mathrm{C} \mathrm{nmr}(100 \mathrm{MHz}$, deuteriochloroform): $\delta-1.7,11.5,11.8,21.3,28.5,51.3,57.3,79.6,105.1$, 204.7, 206.9; ${ }^{77} \mathrm{Se} \mathrm{nmr}$ ( 76 MHz , deuteriochloroform): $\delta 543.5$; $\mathrm{ms}(\mathrm{CI}): \mathrm{m} / \mathrm{z}=304\left(\mathrm{M}^{+}+1\right)$.

Anal. Calcd for $\mathrm{C}_{13} \mathrm{H}_{25} \mathrm{NSe}$ : C, $56.92 ; \mathrm{H}, 9.19 ; \mathrm{N}, 5.11$. Found: C, 56.88; H, 9.12; N, 5.16.

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